
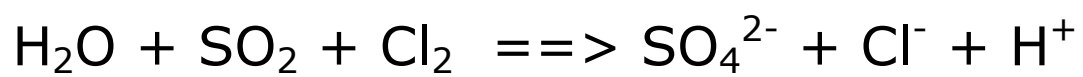


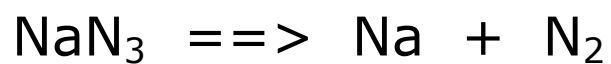
1	$2 \text{CH}_3\text{OH} + 3 \text{O}_2 \implies 2 \text{CO}_2 + 4 \text{H}_2\text{O}$ <p style="text-align: center;">Kode: 12324</p>
2	$\text{C}_4\text{H}_8\text{S} + \text{O}_2 \implies \text{CO}_2 + \text{H}_2\text{O} + \text{SO}_2$ 
3	$\text{C}_3\text{H}_8 + \text{O}_2 \implies \text{CO}_2 + \text{H}_2\text{O}$
4	$\text{CH}_4 + \text{O}_2 \implies \text{CO}_2 + \text{H}_2\text{O}$
5	$\text{AgNO}_3 + \text{Na}_2\text{S} \implies \text{Ag}_2\text{S} + \text{NaNO}_3$
6	$\text{NH}_4\text{NO}_3 \implies \text{N}_2 + \text{O}_2 + \text{H}_2\text{O}$

7	$\text{NH}_4\text{NO}_3 \implies \text{N}_2\text{O} + \text{H}_2\text{O}$
8	$\text{C}_7\text{H}_{16} + \text{O}_2 \implies \text{CO}_2 + \text{H}_2\text{O}$
9	$\text{C}_6\text{H}_{12} + \text{O}_2 \implies \text{CO}_2 + \text{H}_2\text{O}$
10	$\text{H}_2\text{SO}_4 + \text{NaCl} \implies \text{Na}_2\text{SO}_4 + \text{HCl}$
11	$\text{C}_8\text{H}_7\text{N} + \text{O}_2 \implies \text{CO}_2 + \text{H}_2\text{O} + \text{NO}$

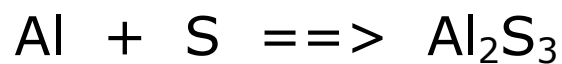
12



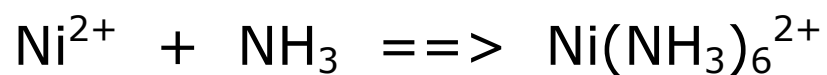
13



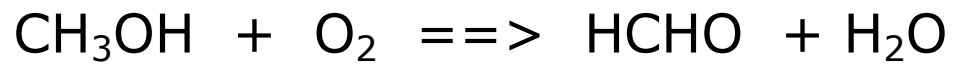
14



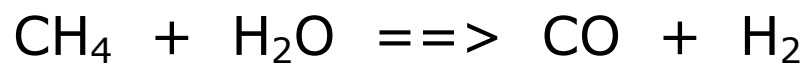
15



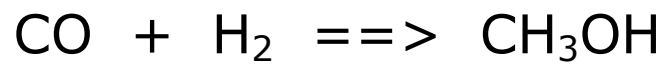
16



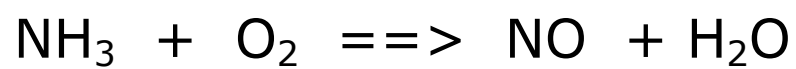
17



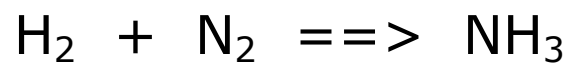
18



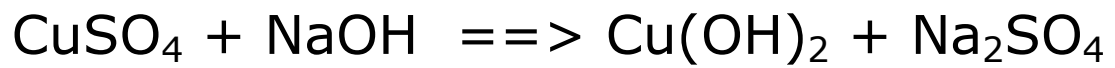
19



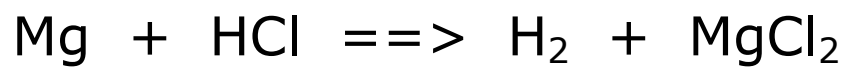
20



21



22



23

